# FIITJEE SAMPLE PAPER (FIITJEE Talent Reward Exam-2020)

for students presently in

Class 9 (Paper 1)



Time: 3 Hours (9:30 am – 12:30 pm)

Code 9000

Maximum Marks: 210

#### Instructions:

Caution: Class, Paper, Code as given above MUST be correctly marked on the answer OMR sheet before attempting the paper. Wrong Class, Paper or Code will give wrong results.

- 1. You are advised to devote 60 Minutes on Section-I, 60 Minutes on Section-II and 60 Minutes on Section-III.
  - Marking Scheme for each question Section Subject Question no. correct answer wrong answer SECTION - I APTITUDE 1 to 30 +3 0 PHYSICS (PART-A) 31 to 39 0 +2 CHEMISTRY 40 to 48 (PART-B) +2 0 SECTION - II MATHEMATICS (PART-C) 49 to 57 +2 0 BIOLOGY (PART-D) 58 to 66 +2 0 PHYSICS (PART-A) 67 to 78 +1 0 CHEMISTRY (PART-B) 79 to 90 +1 0 **SECTION - III** MATHEMATICS (PART-C) 91 to 102 0 +1 BIOLOGY (PART-D) 103 to 114 +1 0
- 2. This Question paper consists of 3 sections. Marking scheme is given in table below:

- 3. Answers have to be marked on the OMR sheet. The Question Paper contains blank spaces for your rough work. No additional sheets will be provided for rough work.
- 4. Blank papers, clip boards, log tables, slide rule, calculator, cellular phones, pagers and electronic devices, in any form, are not allowed.
- 5. Before attempting paper write your OMR Answer Sheet No., Registration Number, Name and Test Centre in the space provided at the bottom of this sheet.

Note: Please check this Question Paper contains all 114 questions in serial order. If not so, exchange for the correct Question Paper.

OMR Answer Sheet No.	:
Registration Number	:
Name of the Candidate	:
Test Centre	:

### Recommended Time: 60 Minutes for Section – I

## Section – I

## APTITUDE TEST

This section contains **30 Multiple Choice Questions** number **1 to 30**. Each question has 4 choices (A), (B), (C) and (D), out of which **ONLY ONE** is correct.

**Directions (Q. 1 to 3):** In each of the following questions, a number/letter series is given with one term missing. Choose the correct alternative that will continue the same pattern and replace the question mark in the given series.

1.	Z, W, S, N, ? (A) P (C) H	(B) O (D) Q
2.	bdf, hjl, ?, tvx. (A) nrp (C) nqr	(B) pnr (D) npr
3.	9, 10, 14, 23, 39, ? (A) 64 (C) 63	(B) 49 (D) 59

4. Find out how many such pairs of letters are there in the given word each of which has as many letters between them in the word as in the English alphabet. ADEQUATELY

(A) One (C) Three			(B) Two (D) Four
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**Directions (Q. 5 to 6):** Read the passage below and solve the questions based on it. Three different faces of a cube are coloured in three different colours – Red, Yellow and Orange. This cube is now cut into 216 smaller but identical cubes.

Space for Rough Work			
11.	Four pairs of words are given out of which the relationship. Choose the pair in which the words (A) Bouquet : Flowers (C) Furniture : Chair	he words in three pairs bear a certain common s are differently related. (B) Bunch : Grapes (D) Album : Photos	
10.	If A + B means A is wife of B; A – B means A Following this relationship, Pankaj – Rajinder = Rahul will certainly mean all of these except (A) Rajinder is a lady (C) Rahul is father of Pankaj	<ul> <li>A is son of B; and A = B means A is sister of B.</li> <li>(B) Pankaj is son of Rajinder</li> <li>(D) None of these</li> </ul>	
9.	Choose the pair/group of words that show the pair/group. Water : Swim (A) Graze : Grass (C) Flood : Damage	<ul> <li>(B) Plan : Implement</li> <li>(D) Ground : Play</li> </ul>	
8.	A word is given in capital letters. It is follower cannot be formed from the letters of the word in formed from the letters of the given word in capit PHILANTHROPIST (A) FIST (C) HYPOCRISY	d by four words. Out of these four words, three n capital letters. Point out the word which can be tal letters. (B) LARK (D) PISTON	
7.	'Melt' is related to 'Liquid' in the same way as 'F (A) Ice (C) Water	reeze' is related to (B) Crystal (D) Cubes	
6.	How many smaller cubes have exactly two face (A) 12 (C) 16	coloured? (B) 15 (D) cannot be determined	
5.	What is the least number of the smaller cubes th (A) 0 (C) 2	nat will have exactly three faces coloured? (B) 6 (D) None of these	

Directions (Q. 12 to 13): According to a certain code,

- (i) 'min fin bin gin' means 'trains are always late';
- (ii) 'gin din cin hin' means 'drivers were always punished':
- (iii) 'min cin vin rin' means 'drivers stopped all trains'; and
- (iv) 'din kin fin vin' means 'all passengers were late'.
- 12. 'Drivers were late' would be written as (B) fin cin din (A) min cin din (C) fin din gin (D) gin hin min
- 13. Which word is represented by 'vin'? (A) all (C) trains
- (B) late (D) drivers
- 14. Seeta and Ram both start from a point towards North. Seeta turns to left after walking 10 km. Ram turns to right after walking the same distance. Seeta waits for some time and then walks another 5 km, whereas Ram walks only 3 km. They both then turn to South and walk 15 km forward. How far is Seeta from Ram? (Akm

(A) 15 km	(B) 10 km
(C) 8 km	(D) 12 km

- 15. X is three times as old as Y, Z was twice as old as X four years ago. In four years time, X will be of 31 years. What is the present age of Y and Z? (A) 9 years, 46 years (B) 9 years, 50 years
  - (C) 10 years, 46 years

- (D) 10 years, 50 years
- 16. Find the missing term in the following figures.



**Directions (Q. 17 to 19):** Read the following information carefully and answer the following questions. Seven person A, B, C, D, E, F and G were born on different months viz. January, February, March, April, June, August and October of the same year but not necessary same order. Only three persons were born before E and D is not one of them. F was not born immediately after E. B was born after F. A was born immediately before the month in which G was born. Only two persons were born between G and F.

17.	How many persons were born between C ar (A) Three (C) Four	nd E? (B) Two (D) Five
18.	Who amongst the following is the oldest?	

Who amongst the following is the oldest?	
(A) A	(B) C
(C) E	(D) B
	Who amongst the following is the oldest? (A) A (C) E

19. Who amongst the following was born between the months in which A and D were born?

(A) F	(B) G
(C) C	(D) B

**Directions (Q. 20 to 22):** Seven persons A, B, C, D, E, F and G are sitting in a circle. Five of them are facing the centre while two of them are facing opposite to the centre. C sits third to the left of D and both are facing the centre. E is neither on immediate neighbour of D nor of C. The one sitting exactly between D and F is facing opposite to the centre. G sits third to right of A and G is facing the centre. One of B's neighbour is facing opposite to the centre.

20.	Which of the following pairs represents persons	facing opposite to the centre.
	(C) A and E	(D) None of these
21	Who is sitting to the left of A2	
21.	(A) C	(B) G
	(C) E	(D) D
22.	Who is sitting to the left of E?	
	(A) C	(B) G
	(C) B	(D) A
~~		

23. Six friends are sitting in a circle and playing cards. Kenny is to the left of Danny. Michael is in between Bobby and Johnny, Roger is in between Kenny and Bobby. Who is sitting to the right of Michael?
 (A) Danny
 (B) Johnny

Danny	
) Kenny	(D) Bobby
	( ) <b>,</b>

(C

**Directions (Q. 24 to 25):** In the following questions, answers are to be based on the diagram given below, where the triangle represents doctors, the circle represents players and the rectangle represent artists.



### Recommended Time: 60 Minutes for Section – II

## Section – II

## PHYSICS - (PART - A)

This part contains **9** Multiple Choice Questions number **31** to **39**. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE is correct.

31.	Relation between mass of body and its weight is		
	(A) w = mg	(B) $w = \frac{m}{q}$	
	(C) $g = m - w$	(D) $w = m + g$	
32.	Which is not the unit of Force : (A) Poundal (C) Joule	(B) Dyne (D) Newton	
33.	The S.I unit of pressure is (A) N.m <sup>2</sup> (C) m <sup>2</sup> /N	(B) N/m <sup>2</sup> (D) N/m	
34.	The momentum of an object at a given instant is (A) inertia (C) velocity	independent of its : (B) speed (D) acceleration	
35.	The atmosphere is held to the earth by (A) Winds (C) Clouds	(B) Gravity (D) None of the above	
Space for Rough Work			

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36.	Value of one Fermi is : (A) $10^{-13}$ meter (C) $10^{-15}$ meter	(B) 10 <sup>-14</sup> meter (D) 10 <sup>-16</sup> meter
37.	Correct relation is (A) $v^2 = u^2 + 2a^2s^2$	(B) $v^2 = u^2 - 2a^2s^2$
	(C) $v^2 = u^2 + 2as$	(D) $v^2 = u^2 + 2a^2 s$

- 38. A bomb at rest and of 10 kg explodes into two pieces of masses 6 kg and 4 kg respectively. The mass of 4 kg moves away with a velocity of 12 ms<sup>-1</sup>. The velocity of the other mass will be:
  (A) 12ms<sup>-1</sup>
  (B) 8ms<sup>-1</sup>
  (C) 4ms<sup>-1</sup>
  (D) 2ms<sup>-1</sup>.
- 39. According to Kepler, the period of revolution of a planet (T) and its mean distance from the sun (r) are related by the equation
  - (A)  $T^3r^3 = constant$ (B)  $T^2r^{-3} = constant$ (C)  $Tr^3 = constant$ (D)  $T^2r = constant$

## CHEMISTRY - (PART - B)

This part contains **9** Multiple Choice Questions number **40** to **48**. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE is correct.

40.	The law of multiple proportions was proposed by (A) Lavoisier (C) Preistley	<ul><li>(B) Dalton</li><li>(D) Ritcher</li></ul>		
41.	180 grams of water contains moles. (A) 100 (C) 180	(B) 10 (D) 0.01		
42.	What is the weight of 3 gram atoms of sulphur? (A) 96 gm (C) 100 gm	(B) 99 gm (D) 3 gm		
43.	The element which is a liquid above 30°C, is : (A) Cesium (C) Sodium	<ul><li>(B) Lithium</li><li>(D) Magnesium</li></ul>		
44.	Separating funnel is useful in separating the follo (A) Miscible liquids with same density (C) Miscible liquids with variable density	owing : (B) Miscible liquids with same colour (D) Immiscible liquids with variable density		
45.	How many moles of oxygen atoms are present i (A) 1 mole (C) 2 moles	n one mole of acetic acid? (B) 3 moles (D) 6 moles		
46.	What is the number of particles in one mole of a (A) $6.023 \times 10^{23}$ (C) $6.023$	substance? (B) $6.023 \times 10^{-23}$ (D) $3 \times 10^{6}$		
47.	The appropriate production of sodium carbonat alcohol is $320 \times 10^6$ g. Which is produced more (A) Sodium carbonate (C) Both (A) and (B)	te per month is $424 \times 10^6$ g while that of methyl in terms of number of moles? (B) Methyl alcohol (D) None of these		
48.	15g of methyl alcohol is present in 100 ml of calculate the mass percentage of methyl alcohol (A) 15.625% (C) 45.625%	solution. If the density of solution is 0.96 g/ml, i n solution. (B) 25.625% (D) 35.625%		
	Space for Rough Work			

## MATHEMATICS – (PART – C)

This part contains **9** Multiple Choice Questions number **49 to 57**. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE is correct.

49.	$0.\overline{37}$ is equivalent to			
	(A) <u>37</u>	(B) <u>55</u>		
	198	(5) 67		
	$(C)\frac{37}{99}$	(D) none of these		
50.	If the number 12x 453 is divisible by 9, then the	digit at the place of x is		
	(A) 1	(B) 2		
	(C) 3	(D) 4		
51.	In the following figure, if AB, CD and EF are straines, find $\angle BOC$ :	aight C F		
	(A) 109 <sup>0</sup> (B) 140 <sup>0</sup>	$A \rightarrow 40^{\circ} O \rightarrow 31^{\circ} B$		
	(C) $71^{\circ}$ (D) $140^{\circ}$	E		
52.	If two interior angles on the same side of a trans ratio 2:3, then the smaller of two angles is:	versal intersecting two parallel lines are in the		
	(A) 72°	(B) 108°		
	(C) 54°	(D) 36°		
53.	The lowest term of $\frac{(x^2-1)(x+2)(x^2-x-72)}{(x-9)(x+1)}$	- is		
	(A) $(x+1)(x-2)(x+8)$	(B) $(x-1)(x+2)(x+8)$		
	(C)(x-1)(x-2)(x+8)	(D) $(x-1)(x+2)(x-8)$		
Space for Rough Work				

54. Find the remainder when  $101\times102\times103\times104\times105\times106\times107$  is divided by 5040 (A) 0 (B) 540 (C) 480 (D) 404

55. The polynomials  $ax^3 + 3x^2 - 13$  and  $2x^3 - 5x + a$  are divided by x + 2. If remainder in each case is the same, the value of a is

- (A)  $\frac{4}{9}$  (B)  $\frac{7}{9}$ (C)  $\frac{2}{9}$  (D)  $\frac{5}{9}$
- 56. Find the LCM of the polynomials:

$90(x^2 - 5x + 6)(2x + 1)^2$ and $140(x - 3)^3$	$^{3}(2x^{2}+15x+7).$
(A) $1260(x-2)(x-3)^{3}(2x+1)^{2}(x+7)$	(B) $1260(x-2)(x+2)^{3}(2x+1)^{2}(x+7)$
(C) $1260(x-2)(x-3)^{3}(2x+1)^{2}(x-7)$	(D) $1260(x-2)(x-3)^{3}(2x-1)^{2}(x+7)$

57. A line segment is of length 10 units. If the co-ordinates of its one end are (2, -3) and the abscissa of the other end is 10, then its ordinate is (A) 9, 6 (B) 3, -9 (C) -3, 9 (D) 9, -6

## BIOLOGY - (PART - D)

This part contains **9** Multiple Choice Questions number **58** to **66**. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE is correct.

58.	The epithelial tissue present on the inner surface (A) Glandular (C) Squamous	e of bronchioles and Fallopian tubes is: (B) Ciliated (D) Cuboidal			
59.	<ul><li>Which one of the following pairs of structures dis</li><li>(A) Vacuoles and fibres</li><li>(C) Nucleus and mitochondria</li></ul>	tinguishes a nerve cell from other types of cell? (B) Flagellum and medullary sheath (D) Perikaryon and dendrites			
60.	<ul><li>Which one of the following contains the largest of</li><li>(A) Striated muscle</li><li>(C) Stratified epithelium</li></ul>	uantity of extracellular material? (B) Areolar tissue (D) Myelinated nerve fibres			
61.	<ul><li>Which one of the following cell organelles is enc</li><li>(A) Lysosomes</li><li>(C) Mitochondria</li></ul>	losed by a single membrane? (B) Nuclei (D) Chlorplasts			
62.	<ul><li>Which of the following structures is not found in</li><li>(A) Mesosome</li><li>(C) Nuclear envelope</li></ul>	a prokaryotic cell? (B) Plasma membrane (D) Ribosome			
<ul> <li>63. The plasma membrane consists mainly of:</li> <li>(A) Phospholipids embedded in a protein bilayer</li> <li>(B) Proteins embedded in a phospholipid bilayer</li> <li>(C) Proteins embedded in a polymer of glucose molecules</li> <li>(D) Proteins embedded in a carbohydrate bilayer</li> </ul>					
64.	<ul><li>Which of the following sets of disease is caused</li><li>(A) Cholera and tetanus</li><li>(C) Tetanus and mumps</li></ul>	by bacteria? (B) Typhoid and smallpox (D) Herpes and influenza			
65.	HIV that causes AIDS, first starts destroying: (A) Helper T-lymphocytes (C) B-lymphocytes	<ul><li>(B) Thrombocytes</li><li>(D) Leucocytes</li></ul>			
66.	The causative agent of mad-cow disease is a: (A) Virus (C) Prion	<ul><li>(B) Bacterium</li><li>(D) Worm</li></ul>			
	Space for Rough Work				

### Recommended Time: 60 Minutes for Section – III

### Section – III

## PHYSICS - (PART - A)

This part contains **12 Multiple Choice Questions** number **67 to 78**. Each question has 4 choices (A), (B), (C) and (D), out of which **ONLY ONE** is correct.

Space for Rough Work				
71.	The period of a satellite in a circular orbit of radi circular orbit of radius 4R is (A) 4 T (C) 8 T	us R is T, the period of another satellite in a (B) T/4 (D) T/8		
70.	The mass of two bodies are 1 kg and 2 respecti respectively. The ratio of their momentum is give (A) 1 : 1 (C) 4 : 1	vely and their kinetic energy are 1 J and 2 J en by. (B) 1 : 4 (D) 1 : 2		
69.	The mass of an object is 10 kg on earth. So we (A) Its weight on earth is 10 N (C) Its weight on moon is 10 N	can say : (B) It weight on Earth is 1.67 N (D) Its mass on moon is 10 kg		
68.	<ul> <li>Value of acceleration due to gravity of earth is n</li> <li>(A) At centre of earth</li> <li>(B) At surface of earth</li> <li>(C) At a height of 50 km from earth's surface</li> <li>(D) At a height of 12 km from earth's surface</li> </ul>	naximum :		
67.	If the mass and radius of earth become half and acceleration due to gravity will be come : (A) $\frac{1}{8}$ g (C) 4 g	(B) $\frac{1}{4}$ g (D) 8 g		
67	If the mass and radius of earth become half and	one-fourth of its present values then he value of		

#### FTRE-2020-C-IX (Paper-1)-AT+S&M-14

72.	Reason of weightlessness in a satellite is (A) Zero gravity (C) Zero reaction force by satellite surface	(B) Centre of mass (D) None
73.	<ul><li>Why we prefer rubber tyres to the steel tyres ?</li><li>(A) Rubber is cheaper than steel</li><li>(B) It is easy to give the rubber a circular shape</li><li>(C) Coefficient of friction between rubber on concrete.</li><li>(D) None of the above</li></ul>	concrete is lower than that between steel on
74.	A block of mass M is placed on a rough floor of and the floor is $\mu$ . When the lift falls freely, the be the force of friction ? (A) $\mu$ Mg (C) $2\mu$ Mg	a lift. The coefficient of friction between the block block is pulled horizontally on the floor. What will (B) $\mu$ Mg/2 (D) None of the above.
75.	A nucleus of mass number (A), originally at rest recoil of the daughter nucleus. (A) $4 v/(A - 4)$ (C) $v/(A - 4)$	emits $\alpha$ particle with speed v. What will be the (B) 4 v/(A + 4) (D) v/(A + 4).
76.	A particle starts its motion from rest under the action of the starts its motion from rest under the action of the start is set of the start in the start is start in the start in the start in the start is start in the start in t	ction of a constant force. If the distanced in first en : (B) $S_2 = 2S_1$ (D) $S_2 = 9S_1$
77.	A stone is dropped from the top of a tower. Its ve $(Take g = 10 m/s^2)$ (A) 5 m/s (C) 30 m/s	elocity after its has fallen 20 m is (B) 10 m/s (D) 20 m/s
78.	When a bus suddenly takes a turn, the passenge (A) Inertia of direction (C) Speed of motion	ers are thrown outwards because of (B) Acceleration of motion (D) Both (B) and (C)

## CHEMISTRY - (PART - B)

This part contains **12 Multiple Choice Questions** number **79 to 90.** Each question has 4 choices (A), (B), (C) and (D), out of which **ONLY ONE** is correct.

79. Chemical analysis of a carbon compound gave the following percentage composition by the elements present in it. Carbon = 10.06%, Hydrogen = 0.84%, Chlorine = 89.10%.					
	(A) $C_2H_2Cl_2$ (C) CHCl <sub>3</sub>	(B) CHCl <sub>2</sub> (D) C <sub>4</sub> H <sub>4</sub> Cl <sub>4</sub>			
80.	Calculate the empirical formula of a compound 26.57, Chromium(Cr)=35.36, Oxygen (O) = $38.0$	having percentage composition: Potassium (K) = $7$			
	(A) $K_2 C C_4$ (C) $K_3 C r_2 O_3$	(b) $K_{2}Cr_{2}O_{7}$ (b) $K_{3}Cr_{2}O_{7}$			
81.	Calculate the mass of 2.5 mole of CaCO <sub>3</sub> . (A) 200 g (C) 240 g	(B) 230 g (D) 250 g			
82.	The purification of drinking water involves : (i) Chlorination (iii) Loading Choose the correct order of these processes (A) i, ii, iii, iiv (C) iv, ii, iii, i	<ul> <li>(ii) Filtration</li> <li>(iv) Sedimentation</li> <li>(B) ii, iv, iii, i</li> <li>(D) iv, iii, ii, i</li> </ul>			
83.	What is the mass of one molecule of oxygen? (A) $5.31 \times 10^{-23}$ g (C) $5.25 \times 10^{23}$ g	(B) $5.33 \times 10^{23}$ g (D) $5.02 \times 10^{-23}$ g			
	Space for Rough Work				

#### FTRE-2020-C-IX (Paper-1)-AT+S&M-16

84. Fractional distillation of two liquids gives better results if the difference is large in their : (B) Densities (A) Boiling points (C) Colours (D) Solubilities 85. Out of two liquids X and Y. X produces more cooling effect than that of Y on the skin. This observation infers that : (A) The boiling point of X is more than that of Y (B) The boiling point of X is less than that of Y (D) The density of X is higher than that of Y (C) The latent heat of X is less than that of Y 86. Calculate the number of atoms present in 6.4 g of sulphur. (A)  $2.4 \times 10^{23}$  atoms (B)  $2.4 \times 10^{-23}$  atoms (C) 1.2 × 10<sup>23</sup> atoms (D)  $1.2 \times 10^{-23}$  atoms 87. The empirical formula of acetic acid is CH<sub>2</sub>O. its molecular weight is 60. Find its molecular formula (A)  $CH_2O$ (B) CH<sub>3</sub>O<sub>2</sub> (C)  $C_2H_4O_2$ (D)  $C_3H_6O_2$ 88. An element has only one type of (A) Molecules (B) Atoms (C) Mixtures (D) Solutes 89. When excess of electrolyte is added to a colloid it (B) Stabilises (A) Coagulates (C) Gets diluted (D) Doesn't change 90. Boot polish contains: (A) Liquid dispersed phase in solid dispersion medium (B) Liquid dispersed phase in liquid dispersion medium (C) Solid dispersed phase in liquid dispersion medium (D) Gas dispersed phase in liquid dispersion medium

## MATHEMATICS - (PART - C)

This part contains **12 Multiple Choice Questions** number **91 to 102**. Each question has 4 choices (A), (B), (C) and (D), out of which **ONLY ONE** is correct.

91.	In triangles ABC and QPR, three equality relatio AB = QP, $\angle B = \angle P$ and BC = PR. State whic (A) SAS (C) SSS	ns between parts are as follows: h of the congruence condition appears? (B) ASA (D) RHS
92.	$ \begin{pmatrix} \frac{a^{m}}{a^{n}} \end{pmatrix}^{m+n} \begin{pmatrix} \frac{a^{n}}{a^{l}} \end{pmatrix}^{n+l} \begin{pmatrix} \frac{a^{l}}{a^{m}} \end{pmatrix}^{l+m} = $ (A) 0 (C) 1/2	(B) 1 (D) -1
93.	If a polynomial, given by $p(x) = k(x-1)(x-2)$ (A) 0 (C) 2	) & $p(0) = 2$ , Then the value of k is (B) 1 (D) None of these.
94.	The sum of all exterior angles of a hexagon is: (A) 180° (C) 360°	(B) 270° (D) 720°
95.	Three or more lines passing through the same p (A) collinear lines (C) concurrent lines	point are called (B) parallel lines (D) coincident lines
96.	The coordinates of one end point of a diameter centre of the circle are $(1, 3)$ . Find the co-ordina (A) $(2, 5)$ (C) $(-2, 5)$	of a circle are $(4, -1)$ and the coordinates of the tes of the other end of the diameter. (B) $(-2, 7)$ (D) $(2, -5)$

97. In  $\triangle ABC$ , if AD divides BC in the ratio m : n, then find the ratio of the areas of  $\triangle ABD \& \triangle ADC$ . (A) m + n : n (B) m : m + n (C) m : n (D) m<sup>2</sup> : n<sup>2</sup>



101. In figure, PQ||RS,  $\angle$ QPR = 70°,  $\angle$ ROT = 20°, find the value of x (A) 20° (B) 70° (C) 110° (D) 50°



102. ABC is an equilateral triangle of side  $4\sqrt{3}$  cm. P, Q and R are midpoints of AB, CA and BC respectively. Find the area of  $\Delta$ PQR

(A) 
$$\frac{\sqrt{3}}{4}$$
 cm<sup>2</sup>  
(B)  $3\sqrt{3}$  cm<sup>2</sup>  
(C)  $2\sqrt{3}$  cm<sup>2</sup>  
(D)  $\frac{\sqrt{3}}{2}$  cm<sup>2</sup>

## BIOLOGY - (PART - D)

This part contains 12 Multiple Choice Questions number 103 to 114. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE is correct.

103. Which type of tissue correctly matches with its location?

	Tissue	Location
(a)	Transitional epithelium	Tip of nose
(b)	Cuboidal epithelium	Lining of stomach
(C)	Smooth muscle	Wall of intestine
(d)	Areolar tissue	Tendons

- 104. Compared to those of humans, the erythrocytes in frog are: (B) Nucleated and with haemoglobin (A) Without nucleus but with haemoglobin (C) Very much smaller and fewer (D) Nucleated and without haemoglobin 105. Areolar connective tissue joins: (A) Bones with bones (B) Fat body with muscles (C) Integument with muscles (D) Bones with muscles 106. Ligament is a/an: (A) Inelastic white fibrous tissue (B) Modified white fibrous tissue
  - (C) Modified yellow elastic fibrous tissue
- (D) None of the above
- Which of the following cell organelles is responsible for extracting energy from carbohydrates to 107. form ATP?
  - (A) Ribosome
  - (C) Mitochondrion

(B) Chloroplast

- (D) Lysosome
- 108. The enzyme recombinase is required at which stage of meiosis?
  - (A) Pachytene (C) Diplotene

(B) Zygotene (D) Diakinesis

109.	During cell growth, DNA synthesis takes place i (A) S-phase (C) G <sub>2</sub> -phase	n: (B) G₁-phase (D) M phase				
110.	<ul> <li>All types of plastids possess essentially the sam</li> <li>(A) Perform the same function</li> <li>(B) Store food materials like starch, fat and prot</li> <li>(C) Occur in aerial parts</li> <li>(D) Can transform from one form to another</li> </ul>	ne structure because they: rein				
111.	Infection of <i>Ascaris</i> usually occurs by: (A) Tse-tse fly (C) Drinking water containing eggs of <i>Ascaris</i>	<ul><li>(B) Mosquito bite</li><li>(D) Eating imperfectly cooked pork</li></ul>				
112.	Identify the site where <i>Wuchereria bancrofti</i> is n (A) Muscles of the legs (C) Skin between the fingers	ormally found in human body. (B) Blood vessels of the thigh region (D) Lymphatic vessels of the lower limbs				
113.	<i>Salmonella</i> is related with: (A) Typhoid (C) T.B.	(B) Polio (D) Tetanus				
114.	The term 'active immunity' means: (A) Increasing rate of heart beat (C) Resistance developed after disease	<ul><li>(B) Increasing quantity of blood</li><li>(D) Resistance developed before disease</li></ul>				
	Space for Rough Work					

Rough

# FIITJEE SAMPLE PAPER – 2020 (FIITJEE Talent Reward Exam-2020)

for students presently in

## Class 9 (Paper 1) ANSWERS

1.	С	:	2. <b>D</b>	3.	Α		4.	С
5.	Α		6. <b>D</b>	7.	Α	3	3.	D
9.	D		10. <b>C</b>	11.	. С	1	12.	в
13.	Α		14. <b>C</b>	15.	. В	1	16.	В
17.	в		18. <b>B</b>	19.	. В	2	20.	С
21.	D	:	22. <b>B</b>	23.	. D	2	24.	В
25.	В	:	26. <b>C</b>	27.	. D	2	28.	Α
29.	С	:	30. <b>D</b>	31.	. Α	3	32.	С
33.	В	:	34. <b>D</b>	35.	. В	3	36.	С
37.	С	:	38. B	39.	. В	4	40.	В
41.	В		42. <b>A</b>	43.	. Α	4	14.	D
45.	С		46. <b>A</b>	47.	. В	4	18.	Α
49.	С		50. <b>C</b>	51.	. В	5	52.	Α
53.	В		54. <b>A</b>	55.	. D	5	56.	Α
57.	В		58. <b>B</b>	59.	. D	6	30.	В
61.	Α		62. <b>C</b>	63.	. В	6	64.	Α
65.	Α		66. <b>C</b>	67.	. D	6	38.	В
69.	D		70. <b>D</b>	71.	. C	7	72.	С
73.	С		74. <b>D</b>	75.	. <b>A</b>	7	76.	D
77.	D		78. <b>A</b>	79.	. C	8	30.	В
81.	D	*	82. <b>D</b>	83.	. <b>A</b>	8	34.	Α
85.	В		86. <b>C</b>	87.	. C	8	38.	В
89.	Α	1	90. <b>A</b>	91.	. <b>A</b>	ę	<del>)</del> 2.	В
93.	В	1	94. <b>C</b>	95.	. C	ç	96.	В
97.	С	1	98. <b>B</b>	99.	. В	1	100.	С
101.	D		102. <b>B</b>	103	3. <b>C</b>	1	104.	В
105.	С		106. <b>C</b>	10	7. <b>C</b>	1	108.	Α
109.	Α		110. <b>D</b>	11	1. <b>C</b>	1	112.	D
113.	Α		114. <b>C</b>					